SYNTHESIS OF BENZO-TYPE PROSTAGLANDIN ANALOGS<sup>1)</sup>

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In recent years, a number of modified prostaglandins(PG) have been synthesized at several laboratories from synthetic and pharmaceutical points of view.<sup>2)</sup> As part of our program of structural modification on PG, attempts have been made to synthesize benzo-type PG derivatives, a new class of PG analogs. We wish to describe here the synthesis of 10,11-benzo-PGA<sub>2</sub> <u>1</u> and its 13,14dihydro derivative 2.





<u>1</u> R= H, <u>17</u> R= CH<sub>3</sub> <u>2</u> R= H, <u>21</u> R= CH<sub>3</sub> Alkylation of indenyllithium with chloromethyl methylether gave 1-methoxymethyl-indene(<u>3</u>, b.p. 50<sup>°</sup>/0.2 mm) in 78% yield<sup>3</sup>), which was lactonized with manganic triacetate<sup>4</sup>) in acetic acid and acetic anhydride at reflux for 0.5 hr to

give the indam lactone<sup>5)</sup> (4, in 68% yield;  $v_{max}$  1775; m/e 218(M<sup>+</sup>)).

The lactone  $\underline{4}$  was reduced with diisobutylaluminium hydride in toluene at -60, and the resulting hemiacetal was condensed with disodium salt of 5-triphenylphosphoniopentanoic acid in dimethyl sulfoxide at room temperature to yield the indanyl hexenoic acid  $\underline{5}$ , which was converted to the methyl ester  $\underline{6}$ .

Direct demethylation of  $\underline{6}$  or its acetate  $\underline{7}$  to  $\underline{12}$  or  $\underline{11}$  failed since undesired dehydration or ring expansion occurred yielding a complex mixture of indene or dihydronaphtalene products. Therefore demethylation with boron tribromide was carried out on indanone ester  $\underline{8}(v_{max} = 1735, 1710, 1605; m/e = 316(M^+))$ , which was obtained from <u>6</u> by oxidation with active manganese dioxide, to obtain the hydroxymethyl indanone 9.

The tetrahydropyranyl ether of <u>9</u> was reduced with sodium borohydride to yield <u>10</u>. Acetylation of <u>10</u>, followed by hydrolysis with aq. acetic acid and tetrahydrofuran afforded the desired hydroxymethyl indanol acetate <u>11</u>(in 68% yield from <u>8</u>;  $v_{max}$  3400,1735). Oxidation of <u>11</u> with Collins reagent in methylene chloride at 0° yielded the aldehyde <u>13</u>, which was condensed with the sodio derivative of dimethyl 2-oxo-heptyl phosphonate in tetrahydrofuran at room temperature to afford the enone <u>14</u>(in 54% yield from <u>11</u>;  $v_{max}$  1740,1695,1630; m/e 440(M<sup>+</sup>), 380(M<sup>+</sup>-60)). Reduction of <u>14</u> with sodium borohydride yielded the alcohol <u>15</u>, which was successively subjected to tetrahydropyranylation and deacetylation by usual procedures to afford <u>16</u>. Oxidation of <u>16</u> with Collins reagent followed by hydrolysis yielded 10,11-benzo-PGA<sub>2</sub> methyl ester <u>17</u>(in 78% yield from <u>14</u>;  $v_{max}$  3400,1740,1715,1605; nmr. & 5.70(2H, m), 5.43(2H, m)). Saponification of <u>17</u> with pottasium carbonate in aq. methanol afforded the objective 10,11-benzo-PGA<sub>2</sub> <u>1</u><sup>6</sup>( $v_{max}$  3400,2650,1710,1605).

On the other hand, 13,14-dihydro-10,11-benzo-PGA<sub>2</sub>  $\underline{2}$  was obtained by the similar procedures described above. The Grignard reaction of 1-(2'-bromoethyl)-indene (b.p. 97°/0.1mm, prepared from indenyllithium and 1,2-dibromoethane) with n-hexanal in ether produced 1-(3'-hydroxyoctyl)-indene <u>18</u> in 57% yield. Treatment of <u>18</u> with manganic triacetate to afford the acetoxylactone <u>19</u>(in 52% yield;  $\nu_{max}$  1780,1735). The lactone <u>19</u> was reduced with disobutylaluminium hydride to the hydroxy-hemiacetal, which was condensed with disodium salt of 5-triphenylphosphoniopentanoic acid to afford the acid <u>20</u>. The methyl ester of <u>20</u> was oxidized with active manganese dioxide to afford the indanone <u>21</u>(in 40% yield from <u>19</u>;  $\nu_{max}$  3400,1735,1605; nmr.  $\delta$  5.37(2H, m); m/e 400(M<sup>+</sup>),382(M<sup>+</sup>-18)), which was saponified with aq. methanol to afford the objective 13,14-dihydro-10,11-benzo-PGA<sub>2</sub> <u>2<sup>6</sup></u>( $\nu_{max}$  3400,2650,1710,1605).

The compounds  $(\underline{1}, \underline{2})$  showed some significant PG like activities such as gastric acid antisecretion or hypotensive activity.











OR	
✓ <sup>1</sup> / <sub>p</sub>	
<sup>·</sup> <sup>R</sup> 1	

<u>10</u>	R= Н,	$R_1 = CH_2OTHP$ ,	$R_2 = CH_3$
<u>11</u>	$R = COCH_3'$	$R_1 = CH_2OH$ ,	$R_2 = CH_3$
<u>12</u>	R= H,	$R_1 = CH_2OH$ ,	$R_2 = CH_3$
<u>13</u>	R= COCH <sub>3</sub> '	$R_1 = CHO,$	$R_2 = CH_3$
<u>14</u>	R= COCH <sub>3</sub> ,	$R_1 = CH = CH = CHCO(CH_2)_4 CH_3'$	$R_2 = CH_3$
15	R= COCH <sub>3</sub> ,	$R_1 = CH=CHCH(CH_2)_4CH_3,$	$R_2 = CH_3$
<u>16</u>	R= H,	$R_1 = CH = CHCH (CH_2) 4^{CH_3}, OTHP$	$R_2 = CH_3$
20	R= H,	$R_{1} = CH_{2}CH_{2}CH_{3}(CH_{2})_{4}CH_{3}$	<sup>R</sup> 2 <sup>= H</sup>



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## References and notes

- Synthetic studies on cyclopentane derivatives Part VIII; proceeding paper of this series. : H. Shimomura, A. Sugie, J. Katsube and H. Yamamoto, <u>Tetrahedron Letters</u>, 4099(1976).
- For examples; P. Crabbe, H. Carpio and A. Guzman, <u>Intra. Sci. Chem. Rep.</u>, <u>6</u>, 55(1972); P. Crabbe, <u>Chem. in Britain</u>, <u>11</u>, 132(1975).
- 3) No amount of the double bond migrated isomer was detected by nmr in this reaction. See; L. Cedheim and L. Eberson, <u>Synthesis</u>, 159(1973); L. Meuring, <u>Acta. Chem. Scand.</u>, 28, 295(1974).
- J.B. Bush and H. Finkbeiner, <u>J. Am. Chem. Soc.</u>, <u>90</u>, 5903(1968); E.J. Heiba,
  R.M. Dessau and W.J. Koehl, <u>ibid</u>., <u>90</u>, 5905(1968).
- 5) The relative configuration between the lactone moiety and the methoxymethyl group is assigned to be trans by following examinations;
  - A. The same lactone was obtained from 4 in following reactions.



B. The vicinal coupling constants observed by spin-decoupling experiments are consistent with the assigned stereochemistry.



6) This product was composed of a mixture of the desired 15-α-OH compound and its 15-epimer (prostanoid numbering), which was subjected to the biological examination without separation. The ratio was found to be almost 1:1 on T.L.C. (Benzene:AcOEt:AcOH = 20:2:1 by 3 times developments); 0.30:0.28 for <u>1</u> and its epimer; 0.29:0.27 for <u>2</u> and its epimer.